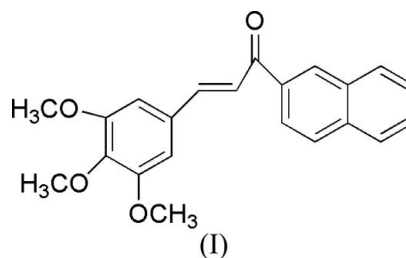


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Key indicators

Single-crystal X-ray study
 $T = 113$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.066
 wR factor = 0.146
Data-to-parameter ratio = 14.5For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**(*E*)-1-(2-Naphthyl)-3-(3,4,5-trimethoxyphenyl)-
prop-2-en-1-one**In the title compound, $\text{C}_{22}\text{H}_{20}\text{O}_4$, the naphthalene ring system makes a dihedral angle of $36.3(5)^\circ$ with the benzene ring. Neighboring molecules are linked *via* weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonding.

Comment

(E)-1-(Naphthalen-2-yl)-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-one, (I), is a valuable pharmaceutical intermediate (Jardine *et al.*, 1980; Issell *et al.*, 1984; Stahelin & Wartburg, 1989). We report here the crystal structure of (I).The molecular structure of (I) is shown in Fig. 1. The dihedral angle between the naphthalene ring system and the benzene ring is $36.3(5)^\circ$. Of the three methoxy groups, two are coplanar with the benzene ring, but the $\text{O}3-\text{C}21$ bond makes an angle of $55.75(13)^\circ$ with the $\text{C}14$ -benzene ring plane, minimizing repulsion between the methoxy groups. Neighboring molecules are linked together *via* weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonding (Table 1).

Experimental

A mixture of 3,4,5-trimethoxybenzaldehyde (5 mmol) and 2-(prop-1-en-2-yl)naphthalenecarbaldehyde (10 mmol) was dissolved in a water-ethanol solution (100 ml, $v/v = 6:4$). A solution of KOH (4.0 ml, 7.1 mmol) was added slowly with stirring at 323 K over 5 h. The resulting mixture was neutralized with 2 *N* HCl solution and extracted with CH_2Cl_2 . The organic layer was washed with brine, dried with Na_2SO_4 and evaporated under vacuum to give (I). Compound (I) (40 mg) was dissolved in a mixture of ethyl acetate (5 ml) and petroleum ether (50 ml), and the solution was allowed to stand at room temperature for 16 d, yielding single crystals of (I).

Crystal data

$\text{C}_{22}\text{H}_{20}\text{O}_4$	$Z = 8$
$M_r = 348.38$	$D_x = 1.307$ Mg m^{-3}
Orthorhombic, $Pbca$	Mo $K\alpha$ radiation
$a = 13.5903(13)$ Å	$\mu = 0.09$ mm^{-1}
$b = 8.6378(9)$ Å	$T = 113(2)$ K
$c = 30.169(3)$ Å	Platelet, colorless
$V = 3541.5(6)$ Å ³	$0.34 \times 0.22 \times 0.06$ mm

Data collection

Rigaku Saturn-70 diffractometer
 ω scans
 Absorption correction: none
 26883 measured reflections

3477 independent reflections
 3229 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.073$
 $\theta_{\text{max}} = 26.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.066$
 $wR(F^2) = 0.146$
 $S = 1.23$
 3477 reflections
 240 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0553P)^2 + 1.8606P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.22 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.21 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.0066 (9)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C12-H12 \cdots O1^i$	0.95	2.56	3.508 (3)	176
$C22-H22B \cdots O1^i$	0.98	2.51	3.405 (3)	153

Symmetry code: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, z$.

Methyl H atoms were placed in calculated positions, with $C-H = 0.98 \text{ \AA}$, and torsion angles were refined to fit the electron density, $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. Aromatic H were placed in calculated positions, with $C-H = 0.93 \text{ \AA}$, and refined in riding mode, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics:

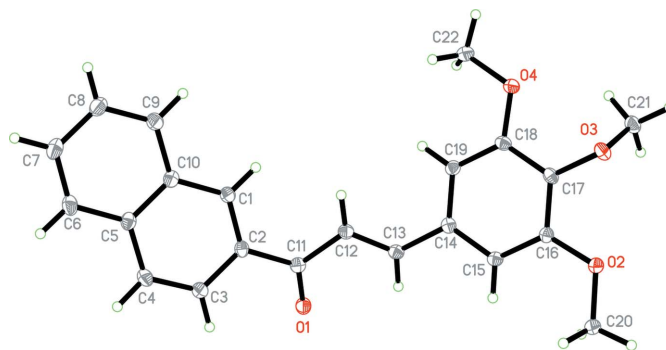


Figure 1

The molecular structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms).

SHELXTL (Bruker, 1997); software used to prepare material for publication: *CrystalStructure* (Rigaku/MS, 2005).

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